

**Trial Examination 2023** 

## **Question and Response Booklet**

# **QCE Physics Units 1&2**

## Paper 1

Student's Name:		
Teacher's Name:		

#### Time allowed

- Perusal time 10 minutes
- Working time 90 minutes

#### **General instructions**

- Answer all questions in this question and response booklet.
- QCAA-approved calculator permitted.
- Formula and data booklet provided.
- Planning paper will not be marked.

#### Section 1 (20 marks)

20 multiple choice questions

#### Section 2 (30 marks)

5 short response questions

#### **SECTION 1**

#### **Instructions**

- Choose the best answer for Questions 1–20.
- This section has 20 questions and is worth 20 marks.
- Use a 2B pencil to fill in the A, B, C or D answer bubble completely.
- If you change your mind or make a mistake, use an eraser to remove your response and fill in the new answer bubble completely.

	A	В	С	D
Example:				

	A	В	С	D
1.	0	$\bigcirc$	0	
2.		$\bigcirc$	$\bigcirc$	$\bigcirc$
3.		$\bigcirc$		$\bigcirc$
4.		$\bigcirc$		$\bigcirc$
5.		00000000000000000		$\bigcirc$
6.		$\bigcirc$	$\bigcirc$	$\bigcirc$
7.		$\bigcirc$	$\bigcirc$	$\bigcirc$
8.		$\bigcirc$		$\bigcirc$
9.		$\bigcirc$	$\bigcirc$	$\bigcirc$
10.		$\bigcirc$	$\bigcirc$	$\bigcirc$
11.		$\bigcirc$	$\circ$	$\bigcirc$
12.		$\bigcirc$		$\bigcirc$
13.		$\bigcirc$	$\bigcirc$	$\bigcirc$
14.		$\bigcirc$	$\bigcirc$	$\bigcirc$
15.		$\bigcirc$	$\bigcirc$	$\bigcirc$
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18.		0 0 0 0	$\bigcirc$	$\bigcirc$
19.		$\bigcirc$	$\bigcirc$	$\bigcirc$
20.		$\bigcirc$	$\bigcirc$	$\bigcirc$

#### **SECTION 2**

#### **Instructions**

- Write using black or blue pen.
- If you need more space for a response, use the additional pages at the back of this booklet.
  - On the additional pages, write the question number you are responding to.
  - Cancel any incorrect response by ruling a single diagonal line through your work.
  - Write the page number of your alternative/additional response, i.e. See page ...
  - If you do not do this, your original response will be marked.
- This section has 5 questions and is worth 30 marks.

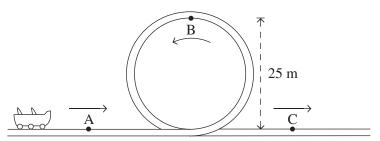
#### DO NOT WRITE ON THIS PAGE

#### THIS PAGE WILL NOT BE MARKED

<b>QUESTION 21</b>	
-	is moving 10 m s <sup>-1</sup> east collides with a 20 kg ball that is moving 20 m s <sup>-1</sup> west. After the s remain in contact with one another.
Determine the fir	nal velocity of the balls.
	Velocity = m s <sup>-1</sup> (to two significant figures)
	Direction =

#### QUESTION 22 (7 marks)

During a safety test, a rollercoaster cart is travelling along a frictionless track. There is a loop in one section of the track, which has a vertical height of 25 metres. As the cart approaches the loop, it passes point A at a speed of  $22.7 \text{ m s}^{-1}$ . As it exits the loop, it passes point C, which is level with point A. This section of track is shown in the diagram.



Not to scale

m s <sup>-1</sup> (to one decimal place)	
oles (that is, joins) with two other carts	
	[2 marks]
	_ m s <sup>-1</sup> (to one decimal place)  ples (that is, joins) with two other carts.

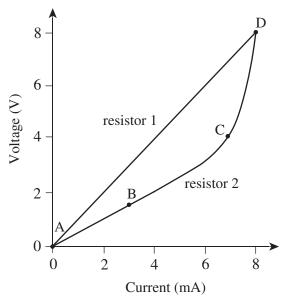
Final speed =

m s<sup>-1</sup> (to one decimal place)

c)	The cart starts the safety test at rest at a location 13 metres before point A.  Assuming uniform acceleration, determine the acceleration of the cart up to point A.	[2 marks]
	$\Delta$ coeleration – $m s^{-2}$ (to the nearest whole number)	

#### QUESTION 23 (4 marks)

Consider the voltage-current graph for two resistors.



a) Explain why resistor 1 is ohmic and resistor 2 is non-ohmic for currents greater than 4 mA.

[2 marks]

b) Determine the value of the resistance on resistor 1.

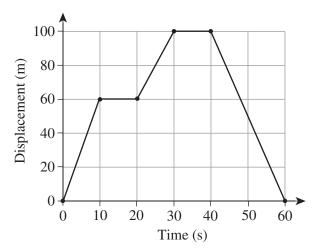
[2 marks]

Resistance =

 $\Omega$  (to one significant figure)

#### QUESTION 24 (7 marks)

The graph shows the change in displacement of a sprinter along a straight running track, beginning at the starting line.



a) For each of the following intervals, describe the motion of the sprinter. Refer to the sprinter's displacement and velocity.

i) 0–10 seconds [1 mark]

ii) 10–20 seconds [1 mark]

iii) 40–60 seconds [1 mark]

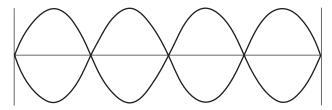
b) Calculate the average velocity for the sprinter's entire journey. [2 marks]

Average velocity = \_\_\_\_\_ m s<sup>-1</sup> (to one significant figure)

Calcu	alculate the average speed for the sprinter's entire journey.		[2 m	
	Average speed -	$m s^{-1}$ (to one significant	t figure)	

## QUESTION 25 (8 marks)

The diagram shows a standing wave within a musical instrument.



Explain whether the instrument is stringed, a pipe open at one end, or a pipe open at both ends.	[3 marks]
Explain the formation of standing waves in terms of superposition. Refer to interference, nodes and antinodes in your response.	[5 marks]

#### **END OF PAPER**

ADDITIONAL PAGE FOR STUDENT RESPONSES	
Write the question number you are responding to.	

ADDITIONAL PAGE FOR STUDENT RESPONSES		
Write the question number you are responding to.		
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**Trial Examination 2023** 

Formula and Data Booklet

# **QCE Physics Units 1&2**

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## **FORMULAS**

Processing of data	
Percentage uncertainty (%) = $\frac{\text{absolute uncertainty}}{\text{measurement}} \times 100$	
Percentage error (%) = $\left  \frac{\text{measured value} - \text{true value}}{\text{true value}} \right  \times 100$	

Heating processes		
$T_{\rm K} = T_{\rm C} + 273$	Q = mL	
$Q = mc\Delta T$	$\Delta U = Q + W$	
$\eta = \frac{\text{energy output}}{\text{energy input}} \times \frac{100}{1}\%$		

Ionising radiation and nuclear reactions	
$N = N_0 \left(\frac{1}{2}\right)^n$	$\Delta E = \Delta m c^2$

Electrical circuits	
$I = \frac{q}{t}$	$P = I^2 R$
$V = \frac{W}{q}$	$V_t = V_1 + V_2 + \dots V_n$
$P = \frac{W}{t}$	$R_t = R_1 + R_2 + \dots R_n$
$R = \frac{V}{I}$	$I_t = I_1 + I_2 + \dots I_n$
P = VI	$\frac{1}{R_t} = \frac{1}{R_1} + \frac{1}{R_2} + \dots + \frac{1}{R_n}$

Linear motion and force	
v = u + at	$W = \Delta E$
$s = ut + \frac{1}{2}at^2$	W = Fs
$v^2 = u^2 + 2as$	$E_{\mathbf{k}} = \frac{1}{2} m v^2$
$a = \frac{F_{\text{net}}}{m}$	$\Delta E_{\rm p} = mg\Delta h$
p = mv	$\sum \frac{1}{2} m v_{\text{before}}^2 = \sum \frac{1}{2} m v_{\text{after}}^2$
$\sum mv_{\text{before}} = \sum mv_{\text{after}}$	

Waves	
$v = f \lambda$	$L = (2n-1)\frac{\lambda}{4}$
$f = \frac{1}{T}$	$\frac{\sin i}{\sin r} = \frac{v_1}{v_2} = \frac{\lambda_1}{\lambda_2} = \frac{n_2}{n_1}$
$L = n\frac{\lambda}{2}$	$I \propto \frac{1}{r^2}$

Gravity and motion	
$v_y = gt + u_y$	$v = \frac{2\pi r}{T}$
$s_y = \frac{1}{2}gt^2 + u_y t$	$a_{\rm C} = \frac{v^2}{r}$
$v_y^2 = 2gs_y + u_y^2$	$F_{\text{net}} = \frac{mv^2}{r}$
$v_x = u_x$	$F = \frac{GMm}{r^2}$
$s_x = u_x t$	$g = \frac{F}{m} = \frac{GM}{r^2}$
$F_g = mg$	$\frac{T^2}{r^3} = \frac{4\pi^2}{GM}$

Electromagnetism	
$F = \frac{1}{4\pi\varepsilon_0} \frac{Qq}{r^2}$	$F = qvB\sin\theta$
$E = \frac{F}{q} = \frac{1}{4\pi\varepsilon_0} \frac{q}{r^2}$	$\phi = BA \cos \theta$
$V = \frac{\Delta U}{q}$	$emf = -\frac{n\Delta(BA_{\perp})}{\Delta t}$
$B = \frac{\mu_0 I}{2\pi r}$	$emf = -n\frac{\Delta\phi}{\Delta t}$
$B = \mu_0 nI$	$I_{p}V_{p} = I_{s}V_{s}$
$F = BIL \sin \theta$	$\frac{V_{\rm p}}{V_{\rm s}} = \frac{n_{\rm p}}{n_{\rm s}}$

Special relativity	
$t = \frac{t_0}{\sqrt{\left(1 - \frac{v^2}{c^2}\right)}}$	$p_{v} = \frac{m_{0}v}{\sqrt{\left(1 - \frac{v^{2}}{c^{2}}\right)}}$
$L = L_0 \sqrt{\left(1 - \frac{v^2}{c^2}\right)}$	$\Delta E = \Delta m c^2$

Quantum theory	
$\lambda_{\max} = \frac{b}{T}$	$\lambda = \frac{h}{p}$
E = hf	$n\lambda = 2\pi r$
$E_k = hf - W$	$mvr = \frac{nh}{2\pi}$
$\frac{1}{\lambda} = R \left( \frac{1}{n_f^2} - \frac{1}{n_i^2} \right)$	

#### PHYSICAL CONSTANTS AND UNIT CONVERSIONS

Heating processes	
Latent heat of fusion for water	$L_{\rm f} = 3.34 \times 10^5  \text{J kg}^{-1}$
Latent heat of vaporisation for water	$L_{\rm v} = 2.26 \times 10^6 \rm J  kg^{-1}$
Specific heat capacity of ice	$c_{\rm i} = 2.05 \times 10^3 \rm J  kg^{-1}  K^{-1}$
Specific heat capacity of steam	$c_{\rm s} = 2.00 \times 10^3 \rm J  kg^{-1}  K^{-1}$
Specific heat capacity of water	$c_{\rm w} = 4.18 \times 10^3 \rm J  kg^{-1}  K^{-1}$

Ionising radiation and nuclear reactions	
Atomic mass unit	1 amu = $1.66 \times 10^{-27}$ kg
Electron volt	$1 \text{ eV} = 1.60 \times 10^{-19} \text{ J}$
Mass of an alpha particle	$m_{\alpha} = 6.6446572 \times 10^{-27} \mathrm{kg}$
Mass of an electron	$m_{\rm e} = 9.1093835 \times 10^{-31} \rm kg$
Mass of a neutron	$m_{\rm n} = 1.6749275 \times 10^{-27} \mathrm{kg}$
Mass of a proton	$m_{\rm p} = 1.6726219 \times 10^{-27} \mathrm{kg}$
Speed of light in a vacuum	$c = 3 \times 10^8 \mathrm{m  s}^{-1}$

Electrical circuits	
Charge on an electron	$e = -1.60 \times 10^{-19} \mathrm{C}$

Linear motion and force	
Mean acceleration due to gravity on Earth	$g = 9.8 \text{ m s}^{-2}$

Waves	
Speed of sound in air at 25°C	$v_{\rm s} = 346 \text{ m s}^{-1}$

Gravity and motion	
Gravitational constant	$G = 6.67 \times 10^{-11} \text{ N m}^2 \text{ kg}^{-2}$
Mass of Earth	$m_{\rm E} = 5.97 \times 10^{24} \rm kg$

Electromagnetism	
Coulomb's constant	$\frac{1}{4\pi\varepsilon_0} = 9 \times 10^9 \text{ N m}^2 \text{ C}^{-2}$
Magnetic constant	$\mu_0 = 4\pi \times 10^{-7} TA^{-1} m$

Quantum theory	
Wien's displacement constant	$b = 2.898 \times 10^{-3} \text{ m K}$
Planck's constant	$h = 6.626 \times 10^{-34} \text{ J s}$
Rydberg's constant	$R = 1.097 \times 10^7 \mathrm{m}^{-1}$

### **SCIENTIFIC NOTATION**

Ratio to basic unit	Prefix	Abbreviation
$10^{-18}$	atto	a
$10^{-15}$	femto	f
$10^{-12}$	pico	p
10 <sup>-9</sup>	nano	n
$   \begin{array}{r}     10^{-9} \\     10^{-6} \\     10^{-3} \\     10^{-2}   \end{array} $	micro	$\mu$
$10^{-3}$	milli	m
$10^{-2}$	centi	c
10 <sup>-1</sup>	deci	d
10	deca	da
10 <sup>2</sup>	hecto	h
10 <sup>3</sup>	kilo	k
10 <sup>6</sup>	mega	M
109	giga	G
10 <sup>12</sup>	tera	Т

### **LIST OF ELEMENTS**

Hydrogen 1 Helium 2 Lithium 3 Beryllium 4 Boron 5 Carbon 6 Nitrogen 7 Oxygen 8 Fluorine 9 Neon 10 Sodium 11 Magnesium 12 Aluminium 13 Silicon 14 Phosphorus 15 Sulfur 16	H He Li Be B C N O F Ne
Lithium 3 Beryllium 4 Boron 5 Carbon 6 Nitrogen 7 Oxygen 8 Fluorine 9 Neon 10 Sodium 11 Magnesium 12 Aluminium 13 Silicon 14 Phosphorus 15 Sulfur 16	Li Be B C N O F Ne
Beryllium         4           Boron         5           Carbon         6           Nitrogen         7           Oxygen         8           Fluorine         9           Neon         10           Sodium         11           Magnesium         12           Aluminium         13           Silicon         14           Phosphorus         15           Sulfur         16	Be B C N O F Ne
Boron         5           Carbon         6           Nitrogen         7           Oxygen         8           Fluorine         9           Neon         10           Sodium         11           Magnesium         12           Aluminium         13           Silicon         14           Phosphorus         15           Sulfur         16	B C N O F
Carbon6Nitrogen7Oxygen8Fluorine9Neon10Sodium11Magnesium12Aluminium13Silicon14Phosphorus15Sulfur16	C N O F Ne
Nitrogen7Oxygen8Fluorine9Neon10Sodium11Magnesium12Aluminium13Silicon14Phosphorus15Sulfur16	N O F Ne
Oxygen 8 Fluorine 9 Neon 10 Sodium 11 Magnesium 12 Aluminium 13 Silicon 14 Phosphorus 15 Sulfur 16	O F Ne
Fluorine 9 Neon 10 Sodium 11 Magnesium 12 Aluminium 13 Silicon 14 Phosphorus 15 Sulfur 16	F Ne
Neon10Sodium11Magnesium12Aluminium13Silicon14Phosphorus15Sulfur16	Ne
Sodium11Magnesium12Aluminium13Silicon14Phosphorus15Sulfur16	
Magnesium12Aluminium13Silicon14Phosphorus15Sulfur16	
Aluminium 13 Silicon 14 Phosphorus 15 Sulfur 16	Na
Silicon 14 Phosphorus 15 Sulfur 16	Mg
Phosphorus 15 Sulfur 16	Al
Sulfur 16	Si
	P
	S
Chlorine 17	Cl
Argon 18	Ar
Potassium 19	K
Calcium 20	Ca
Scandium 21	Sc
Titanium 22	Ti
Vanadium 23	V
Chromium 24	Cr
Manganese 25	Mn
Iron 26	Fe
Cobalt 27	Со
Nickel 28	Ni
Copper 29	Cu
Zinc 30	Zn
Gallium 31	Ga
Germanium 32	Ge
Arsenic 33	As
Selenium 34	
Bromine 35	Se

Name	Atomic no.	Symbol							
Krypton	36	Kr							
Rubidium	37	Rb							
Strontium	38	Sr							
Yttrium	39	Y							
Zirconium	40	Zr							
Niobium	41	Nb							
Molybdenum	42	Мо							
Technetium	43	Тс							
Ruthenium	44	Ru							
Rhodium	45	Rh							
Palladium	46	Pd							
Silver	47	Ag							
Cadmium	48	Cd							
Indium	49	In							
Tin	50	Sn							
Antimony	51	Sb							
Tellerium	52	Те							
Iodine	53	I							
Xenon	54	Xe							
Cesium	55	Cs							
Barium	56	Ba							
Lanthanum	57	La							
Cerium	58	Ce							
Praseodymium	59	Pr							
Neodymium	60	Nd							
Promethium	61	Pm							
Samarium	62	Sm							
Europium	63	Eu							
Gadolinium	64	Gd							
Terbium	65	Tb							
Dysprosium	66	Dy							
Holmium	67	Но							
Erbium	68	Er							
Thulium	69	Tm							
Ytterbium	70	Yb							

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## LIST OF ELEMENTS (CONTINUED)

Name	Atomic no.	Symbol							
Lutetium	71	Lu							
Hafnium	72	Hf							
Tantalum	73	Та							
Tungsten	74	W							
Rhenium	75	Re							
Osmium	76	Os							
Iridium	77	Ir							
Platinum	78	Pt							
Gold	79	Au							
Mercury	80	Hg							
Thallium	81	Tl							
Lead	82	Pb							
Bismuth	83	Bi							
Polonium	84	Po							
Astatine	85	At							
Radon	86	Rn							
Francium	87	Fr							
Radium	88	Ra							
Actinium	89	Ac							
Thorium	90	Th							
Protactinium	91	Pa							
Uranium	92	U							
Neptunium	93	Np							
Plutonium	94	Pu							

Name	Atomic no.	Symbol							
Americium	95	Am							
Curium	96	Cm							
Berkelium	97	Bk							
Californium	98	Cf							
Einsteinium	99	Es							
Fermium	100	Fm							
Mendelevium	101	Md							
Nobelium	102	No							
Lawrencium	103	Lr							
Rutherfordium	104	Rf							
Dubnium	105	Db							
Seaborgium	106	Sg							
Bohrium	107	Bh							
Hassium	108	Hs							
Meitnerium	109	Mt							
Darmstadtium	110	Ds							
Roentgenium	111	Rg							
Copernicium	112	Cn							
Nihonium	113	Nh							
Flerovium	114	Fl							
Moscovium	115	Мс							
Livermorium	116	Lv							
Tennessine	117	Ts							
Oganesson	118	Og							

6		10			9			36			24		6	98		(C	118					11				103		_
18	<b>He</b>		Se	20.18		Ar	39.95		7	83.80		Xe	131.29		R	(222.0)	_	0g	(294)		ı		P.	174.97			Ļ	(262.1)
	17	6	щ	19.00	17	5	35.45	35	Br	79.90	53	_	126.90	85	At	(210.0)	117	Ls	(294)		ı	70	Λþ	173.05		102	2 S	(259.1)
	16	8	0	16.00	16	S	32.06	34	Se	78.97	52	Te	127.60	84	Po	(210.0)	116	<b>_</b> ^	(293)			69	E	168.93		101	Β	(258.1)
	15	7	2	14.01	15	Д	30.97	33	As	74.92	51	Sb	121.76	83	<u>.</u>	208.98	115	Mc	(288)			89	ъ	167.26		100	Fm	(252.1)
	14	9	ပ	12.01	14	Si	28.09	32	Ge	72.63	20	Sn	118.71	82	Pp	207.2	114	Ξ	(289)			29	<b>9</b>	164.93		66	Es	(252.1)
	13	5	<b>m</b>	10.81	13	A	26.98	31	Ga	69.72	49	므	114.82	81	F	204.38	113	Z	(284)			99	Ď	162.50		98	Ç	(252.1)
							12	30	Zn	65.38	48	Cq	112.41	80	Hg	200.59	112	Cu	(282)			69	Q L	158.93		97	B	(249.1)
TS							11	29	Cn	63.55	47	Ag	107.87	79	Au	196.97	111	Rg	(272)			64	P9	157.25		96	Cm	(244.1)
ELEMEN							10	28	Z	58.69	46	Pd	106.42	78	Ŧ	195.08	110	Ds	(281)			63	ш	151.96		95	Am	(241.1)
E OF THE	į	umber	symbol relative atomic mass*				6	27	Ç	58.93	45	Rh	102.91	77	<u>_</u>	192.22	109	Ĭ	(268)			62	Sm	150.36		94	Pu	(239.1)
DIC TABLE OF THE ELEMENTS	_	1 atomic number	symbol relative a				80	26	Fe	55.85	44	Ru	101.07	9/	0s	190.23	108	Hs	(265.1)		ı	61	Pm	(146.9)		93	Νp	(237.0)
PERIOD	KEY		<b>L</b> (1				7	25	Ξ	54.94	43	٦ <sub>C</sub>	(98.91)	75	Re	186.21	107	Bh	(264.1)			09	Z	144.24		92	<b>-</b>	238.0
							9	24	ت	52.00	42	Θ	95.95	74	>	183.84	106	Sg	(263.1)			59	Pr	140.91		91	Pa	231.0
							2	23	>	50.94	41	S	92.91	73	Та	180.95			(262.1)		ı	28	Çe	140.12		90	T L	232.0
							4	22	ï	47.87	40	Zr	91.22	72	¥	178.49	104	Rf	(261.1)		Lanthanoids	22	Га	138.91	Actinoids	88	Ac	(227.0)
							က	21	Sc	44.96	39	>	88.91	57-71	Lanthanoids		89-103	Actinoids	_		اتـ 		^    - +  -				<u></u>	
	2	4	Be	9.01	12	Σ	24.31	20	Ca	40.08	38	Sr	87.62	99	Ва	137.33	88	Ba	(226.1)	_								
	<b>T</b> 10:	က	=	6.94	1	Na	22.99	19	¥	39.10	37	Rb	85.47	22	Cs	132.91	87	Ŧ	(223.0)									

Groups are numbered according to IUPAC convention 1–18. \*Values in brackets are for the isotope with the longest half-life.