# **PEARSON**

# **Year 12 Physics**

**Practice Exam 1 (Units 3 & 4) Answers**

**Unit 3**

**Area of Study 1 How do things move without contact?**

#### **Question 1**



### **Question 2**



#### **Question 3**

 $v = 1.9 \times 10^5$  m s<sup>-1</sup>



m s**−1** (1 mark)

$$
4.8 \times 10^{-3} = \frac{0.67 \times 10^{-14} \text{ m}}{(14 \times 10^{3} + 1.2)^{2}}
$$
 (1 mark)  
\n
$$
M = 1.4 \times 10^{16} \text{ kg}
$$
 (1 mark)

### **Question 4**

 $\frac{T^2}{r^3} = \frac{4\pi^2}{GM}$ *GM*  $r^3 = \frac{GMT^2}{4\pi^2}$ <sup>4</sup>*π***<sup>2</sup>** (1 mark)  $r^3 = \frac{6.67 \times 10^{-11} \times 1.99 \times 10^{30} \times (410 \times 24 \times 3600)^2}{4 \pi^2}$  $4\pi^2$ *r***<sup>3</sup>** = 4.22 × 10**<sup>33</sup>**  $r = 1.62 \times 10^{11} \text{ m}$  (2 marks)



# **Area of Study 2 How are fields used to move electrical energy?**

## **Question 6**





## **Question 7**







**a** Two arrows, weight acting vertically down and normal force acting at right angles to the surface angles to the surface of the  $(1 \text{ mark})$ 





**c**

**d**











**b** 
$$
E_s = \frac{1}{2}k\Delta x^2
$$
  
\n $E_s = \frac{1}{2} \times 98 \times 0.2^2$  (1 mark)  
\n**c**  $E_s = \frac{1}{2}k\Delta x^2$   
\n $E_s = \frac{1}{2} \times 98 \times 0.3^2$   
\n $E_s = 4.41$  J (1 mark)  
\n**d** total energy measured at  $x = 0.3$  m = 4.41 J (1 mark)  
\nAt B: 4.41 =  $mg\Delta h + \frac{1}{2}mv^2 + \frac{1}{2}k\Delta x^2$   
\n4.41 = 2 × 9.8 × 0.1 +  $\frac{1}{2} \times 2 \times v^2 + \frac{1}{2} \times 98 \times 0.2^2$   
\n $\frac{1}{2} \times 2 \times v^2 = 4.41 - 1.96 - 1.96$  (1 mark)

$$
v = 0.7 \text{ m s}^{-1}
$$
 (1 mark)

$$
L = \frac{L_0}{\gamma} = L_0 \times \sqrt{1 - \frac{v^2}{c^2}}
$$
  
\n
$$
60 = 75 \times \sqrt{1 - \frac{v^2}{c^2}}
$$
  
\n
$$
\frac{60}{75} = \sqrt{1 - \frac{v^2}{c^2}}
$$
  
\n
$$
(\frac{60}{75})^2 = 1 - \frac{v^2}{c^2}
$$
  
\n
$$
v^2 = 0.36c^2
$$
  
\n
$$
v = 0.6c
$$
  
\n(1 mark)

## **Unit 4**

## **Area of Study 1 How can waves explain the behaviour of light?**

## **Question 14**

**a** coherent light, e.g., laser light (1 mark)

Incoherent light will not result in the interference pattern observed by Young. There will not be the pattern of constructive and destructive interference that produces the bright and dark bands. (1 mark)

**b** (2 marks)





## **Area of Study 2 How are light and matter similar?**

## **Question 17**



#### **Question 18**

- **a** The single hydrogen electron is in the ground state, so it has energy of 13.6 eV. (1 mark) The photon energy is greater than 13.6 eV, so the electron will acquire sufficient energy to be ejected from the atom. (1 mark)
- **b** energy of ejected electron =  $14.0 13.6 = 0.4$  eV (1 mark) in joules, *E* = 0.4 × 1.6 × 10**−19** = 6.40 × 10**−20** J (1 mark)

#### **Question 19**

**a** 
$$
\lambda = \frac{h}{mv}
$$
  
\n
$$
\lambda = \frac{6.63 \times 10^{-34}}{9.1 \times 10^{-31} \times 1.8 \times 10^7}
$$
  
\n
$$
\lambda = 4.05 \times 10^{-11} \text{ m}
$$
 (1 mark)  
\n**b** The radius of the diffraction pattern would decrease. (1 mark)  
\nAs  $\lambda = \frac{h}{mv}$  an increase in velocity will reduce the wavelength. (1 mark)  
\nAs the amount of diffraction is proportional to the wavelength, decreasing  $\lambda$  will result in the radius of the diffraction pattern getting smaller. (1 mark)  
\n**c** 
$$
E = \frac{hc}{\lambda}
$$
  
\n
$$
E = \frac{6.63 \times 10^{-34} \times 3 \times 10^8}{4.05 \times 10^{-11}}
$$
  
\n
$$
E = 4.91 \times 10^{-15} \text{ J}
$$
 (1 mark)

**d** Diffraction is a property of wave-like behaviour. **Example 20 and Conventional** (1 mark)

Thus, as the electrons have exhibited a diffraction pattern, they must have wave-like properties. (1 mark)



- **b** 3.5 eV, from extending the line of best fit back to the (vertical) energy axis (1 mark)
- **c** According to the wave model, a threshold frequency for the incident photons would not be required. This is because the wave model predicts that all the incident photons would be absorbed, resulting in the electrons eventually gaining enough energy to escape the metal. This behaviour is not observed (1 mark)

The particle model correctly predicts that unless an incident photon has sufficient energy to enable an electron to escape the metal, the photon will not be absorbed. The threshold frequency is the minimum frequency for which photons will have enough energy to liberate an electron. (1 mark)

**d** 
$$
E = \frac{hc}{\lambda}
$$
  
\n $E = \frac{4.38 \times 10^{-15} \times 3 \times 10^8}{460 \times 10^{-9}}$  (1 mark)  
\n $E = 2.86 \text{ eV}$  (1 mark)  
\n $E_k = 0.76 \text{ eV} = 1.22 \times 10^{-19}$  (1 mark)  
\n $E_k = \frac{1}{2} m v^2$   
\n $v = \sqrt{\frac{2E_k}{m}} = \sqrt{\frac{2 \times 1.22 \times 10^{-19}}{9.1 \times 10^{-31}}}$  (1 mark)  
\n $v = 5.18 \times 105 \text{ m s}^{-1}$  (1 mark)  
\nCs line drawn parallel to Sc line  
\n(1 mark)  
\nCs line drawn to give energy axis intercept of 2.1 eV (1 mark)